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EDITORIAL



Is cytochrome oxidase inhibition the primary mechanism in aluminum phosphide poisoning?

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Aluminum phosphide (AIP) has common applications as rodenticides and grain fumigants in Asian countries. It has a very high mortality rate [1–3]. Overall mortality related to AIP reported from 30% to 100% [1–3]. The cause of which is a cardiogenic shock [4]; it also induces severe metabolic acidosis and refractory hypotension [5]. Although it is a significant concern in many countries, the pathophysiological mechanisms have not been clearly explained in humans. However, several articles suggest that it can inhibit cytochrome c oxidase activity in severely poisoned patients [1,5–8]. Oxidative stress is another suggested mechanism [1,8].

Moreover, complexation of calcium and magnesium is another suggested mechanism of toxicity induced by AIP. Both hypomagnesemia and hypermagnesemia are reported in AIP poisoning [9,10].

In a previous study, Anand et al. evaluated mitochondrial electron transport chain complexes, oxidative stress, and catalase markers in patients with severe AIP poisoning. They concluded that AIP not only inhibits cytochrome c oxidase but also affects other enzymes, resulting in severe energy insufficiency and increased production of reactive oxygen species, which can synergistically cause tissue damage [8]. Moreover, Dua et al. found that the interaction of AIP with redox chain components causes the impairment of the electron transfer along the respiratory chain [11].

In fact, in the literature, inhibition of cytochrome c oxidase is a suggested mechanism for the toxicity of AIP. Cytochrome oxidase, which is recognized as an iron-containing enzyme, is necessary for oxidative phosphorylation and, therefore, aerobic energy generation. This enzyme acts in the mitochondrial electron transport chain, converting catabolic glucose products into adenosine triphosphate (ATP).

There are some major concerns in this area:

There are some similarities between AIP, cyanide, hydrogen sulfide, and azide because all of these toxins induce cellular hypoxia through inhibition of cytochrome oxidase and electron transport chain. In cyanide, hydrogen sulfide, and azide poisoning, oxygen cannot be used despite adequate oxygen supply; therefore, ATP molecules cannot be generated. Unincorporated hydrogen ions accumulate, leading to metabolic acidosis. Because of failure in aerobic energy metabolism, hyperlactatemia may ensue, especially in cyanide poisoning [12]. In general,

the mechanisms of hyperlactatemia are cellular hypoxia due to hypoperfusion, hyperstimulation of Na^+/K^+ -ATPase, increased pyruvate and lactate because of rise in anaerobic glycolysis, lactate clearance decline, seizures inducing muscle hyperactivity, and electron transfer and oxidative phosphorylation impairment. Baud et al. recently revealed that hyperlactatemia as the single factor has an ability to predict cyanide poisoning [13]. Moreover, they found that the level of blood cyanide meaningfully associated with the blood lactate increasing. Interestingly in cases with lactic acidosis, the median serum creatinine phosphokinase activity was not significantly different from that in control patients [13].

There are very few articles evaluating lactate levels in cases of AIP poisoning. Erfantalab et al. investigated the trend of lactate in this poisoning; they found that there is a short time hyperlactatemia at the initial phase of poisoning, although the trend was descending, and it returned to the normal range at the end of the first day of admission [14]. In addition, because of reduced oxygen utilization in tissues, high venous oxygen saturation is usually reported in these cases of poisoning.

Although the suggested mechanism of AIP poisoning is very similar to the abovementioned poisoning, elevated venous oxygen saturation is rarely reported following AIP poisoning. Moreover, hyperlactatemia may be seen in all of these poisoning, but the trend is different between them.

In addition, one of the consequences of cellular energy depletion is rhabdomyolysis. However, this is rarely reported following AIP poisoning. Besides, coma is a major finding induced by inhibitors of cytochrome oxidase, including cyanide, hydrogen sulfide, and carbon monoxide but was not reported in AIP poisoning. In fact, patients may remain conscious until the late stage. Coma may happen as a result of anoxia due to refractory shock, not due to the neurotoxicity of AIP [15].

If cytochrome c oxidase inhibition is a major factor in AIP toxicity, rhabdomyolysis, high venous oxygen saturation, and coma in poisoned patients may be expected. However, this finding is rarely reported. Although we should be aware that AIP occurs far less frequently and often in developing countries that may not evaluate for or report arterial blood gas/venous blood gas mismatch and rhabdomyolysis.

In conclusion, there are few pieces of evidence against the role of cytochrome c oxidase inhibition in the literature. Also, it seems

that some aspects of AIP poisoning are not evaluated. So, we strongly encourage researchers to evaluate the other mechanisms in this lethal poisoning. The knowledge of a complex mechanism of acute toxicity of AIP is important and necessary to reach a precise diagnosis of poisoning and to develop adequate treatment of acute poisoning with AIP. We encourage researchers to evaluate the exact mechanism of AIP-induced toxicity.

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