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SHORT COMMUNICATION

Inhibition of the β -class carbonic anhydrases from *Mycobacterium tuberculosis* with carboxylic acids

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Abstract

The growth of *Mycobacterium tuberculosis* is strongly inhibited by weak acids although the mechanism by which these compounds act is not completely understood. A series of substituted benzoic acids, nipecotic acid, *ortho*- and *para*-coumaric acid, caffeic acid and ferulic acid were investigated as inhibitors of three β -class carbonic anhydrases (CAs, EC 4.2.1.1) from this pathogen, mtCA 1 (Rv1284), mtCA 2 (Rv3588c) and mtCA 3 (Rv3273). All three enzymes were inhibited with efficacies between the submicromolar to the micromolar one, depending on the scaffold present in the carboxylic acid. mtCA 3 was the isoform mostly inhibited by these compounds (K_i s in the range of 0.11–0.97 μ M); followed by mtCA 2 (K_i s in the range of 0.59–8.10 μ M), whereas against mtCA 1, these carboxylic acids showed inhibition constants in the range of 2.25–7.13 μ M. This class of relatively underexplored β -CA inhibitors warrant further *in vivo* studies, as they may have the potential for developing antimycobacterial agents with a diverse mechanism of action compared to the clinically used drugs for which many strains exhibit multi-drug or extensive multi-drug resistance.

Keywords: Carbonic anhydrase, β -carbonic anhydrase, carboxylic acid, benzoic acid, coumaric acid, ferulic acid, enzyme inhibitor, *Mycobacterium tuberculosis*

Introduction

Mycobacterium tuberculosis has been found to be highly sensitive to weak organic acids, (most of which incorporate aromatic scaffolds) administered as amide-type prodrugs¹. Indeed, two first line medications for treating tuberculosis are constituted by hydrazides/amides of two carboxylic acid, i.e. isoniazide (isonicotinoyl hydrazide **A**) and pyrazinamide **B**¹. Carboxylic acids on the other hand also act as inhibitors of the zinc enzyme carbonic anhydrase (CA, EC 4.2.1.1)^{2–5}. Indeed, simple carboxylates such as formate and acetate were investigated as mammalian CA inhibitors (CAIs) by Scozzafava's group who showed, by using paramagnetic NMR and UV-VIS spectroscopy of Co(II)-substituted CAs complexed with these anions, the metal ion is pentacoordinated in these adducts by three His residues from the enzyme, a water molecule and a carboxylate anion from the inhibitor^{6,7}. This has been thereafter confirmed by X-ray

crystallographic studies which showed these anions to bind to the metal ion in a monocoordinated manner with one oxygen of the carboxylate moiety bound to the Zn(II) ion, which is indeed in a trigonal bipyramidal geometry, since a water molecule is also coordinated to it (in addition to the carboxylate ligand)⁸. In contrast to the α -class CAs present in mammals, including humans^{2–5}, many β -CAs are found in bacterial/fungal pathogens and their inhibition started to be investigated only recently as a possibility to obtain novel types of antibiotics/antifungals^{9,10}. For example, bacterial enzymes from *Escherichia coli*, *Helicobacter pylori*, *M. tuberculosis*, *Brucella spp.*, *Streptomyces pneumoniae*, *Salmonella enterica* and *Haemophilus influenzae* were cloned and characterized in the last years⁹. The catalytic/inhibition mechanisms of these CAs are well understood, as X-ray crystal structures are available for some of them, but no adducts of these enzymes with inhibitors were characterized so

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far by this technique, except the acetate adduct of an enzyme (Can2) from the fungal pathogen *Cryptococcus neoformans*¹⁰. As for the α -class CAs mentioned above^{6–8}, acetate is coordinated to the Zn(II) ion from the Can2 active site, substituting the water molecule/hydroxide ion responsible for the catalytic activity. Thus, Zn(II) is in a tetrahedral geometry in this complex¹⁰.

The genome of the human pathogen *M. tuberculosis* encodes at least three β -CAs (EC 4.2.1.1), called mtCA 1, 2 and 3, which are encoded by the genes Rv1284, Rv3588c Rv3273^{11–16}. These enzymes have been cloned and their catalytic activity investigated^{11–16}. As most CAs described to date, these enzymes are inhibited by sulfonamides and related compounds^{17–23}. However, although interesting, low nanomolar or subnanomolar sulfonamide inhibitors were detected^{3–6}, it is not yet clear whether their *in vivo* inhibition has an antimycobacterial effect¹⁷.

Resistance to antibiotics belonging to several different classes is escalating and represents a worldwide problem^{24–28}. Many strains of Gram-negative/-positive bacteria (such as among others *Staphylococcus aureus*, *M. tuberculosis*, *H. pylori*, *B. suis*, *S. pneumoniae*, etc.) no longer respond to various classes of antibiotics^{24–28}. The situation is particularly dramatic for *M. tuberculosis* with the emergence of multi-drug resistant and extensive multi-drug resistant strain all over the world^{24–28}. It is thus stringent to investigate alternative mycobacterial targets as well as different chemotypes (than the usual ones for these targets) in order to identify novel lead compounds which may help the development of efficient antibiotics.

The classical CA inhibitors (CAIs) are the primary sulfonamides, RSO_2NH_2 , which are used clinically for more than 50 years as diuretics or systemically acting anti-glaucoma drugs^{2–5}. More than 30 clinically used drugs (or compounds in clinical development) belonging to the sulfonamide or sulfamate class, show significant CA inhibitory activity and are used for the management of a variety of disorders^{2–5}. However, all these drugs target all mammalian CAs, of which 16 different isoforms are known so far, and as thus, they may show undesired side effects^{2–5}. Many of them also inhibit efficiently the bacterial/fungal β -CAs⁹. As thus, as mentioned above, a large number of such derivatives started to be investigated as possible anti-infectives with the aim of specifically targeting bacterial/fungal CAs. This field is however still a new one with relevant resistance from the scientific community (not only from bacteria and fungi) in accepting these enzymes as anti-infective targets⁹. It should be also mentioned that sulfonamides, of which the prototypical drug is the bacteriostatic drug sulfanilamide **C**, do not show antimycobacterial effects¹⁷ (Figure 1).

Thus, exploring various chemotypes, other than the sulfonamide one, may lead to the discovery of β -CA inhibitors with good affinity for enzymes from pathogenic species. Based on the fact that carboxylic acids are reported to show relevant antimycobacterial effects¹ and that many such compounds act as CAIs^{5–9}, here we investigated a

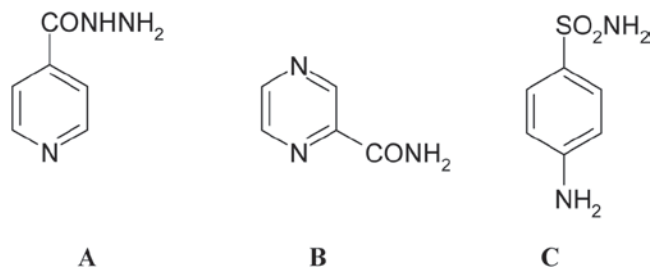


Figure 1. Structures of compounds isoniazide, pyrazinamide and sulfanilamide.

series of carboxylic acids incorporating various aromatic/heterocyclic scaffold as inhibitors of the three β -CAs from *M. tuberculosis*, i.e. mtCA 1 – mtCA 3, evidencing interesting biological activities for many of them.

Materials and methods

Chemistry

Carboxylic acids **1–10** investigated here were commercially available from Sigma-Aldrich, except *o*-coumaric acid (sodium salt) which was prepared by the procedure of Maresca et al.²⁹ from coumarin by alkaline hydrolysis.

Enzymology

mtCA1–mtCA 3 were recombinant enzymes obtained as described earlier^{11–17}.

CA catalytic activity and inhibition assay

An Applied Photophysics stopped-flow instrument has been used for assaying the CA-catalyzed CO_2 hydration activity³⁰. Phenol red (at a concentration of 0.2 mM) has been used as indicator, working at the absorbance maximum of 557 nm with 10–20 mM Hepes (pH 7.5, for α -CAs) or TRIS (pH 8.3 for β -CAs) as buffers, and 20 mM Na_2SO_4 (for α -CAs) or 20 mM NaCl for β -CAs (for maintaining constant the ionic strength), following the initial rates of the CA-catalyzed CO_2 hydration reaction for a period of 10–100 s. The CO_2 concentrations ranged from 1.7 to 17 mM for the determination of the kinetic parameters and inhibition constants. For each inhibitor, at least six traces of the initial 5–10% of the reaction have been used for determining the initial velocity. The uncatalyzed rates were determined in the same manner and subtracted from the total observed rates. Stock solutions of inhibitor (10 mM) were prepared in distilled-deionized water and dilutions up to 0.01 nM were done thereafter with distilled-deionized water. Inhibitor and enzyme solutions were preincubated together for 15 min at room temperature prior to assay in order to allow for the formation of the E-I complex. The inhibition constants were obtained by non-linear least-squares methods using PRISM 3, whereas the kinetic parameters for the uninhibited enzymes from Lineweaver-Burk plots, as reported earlier^{11–17,31}, and represent the mean from at least three different determinations.

Results and discussions

We report here the investigation of a series of carboxylic acids of type **1–10** for their interaction with three β -CAs from the bacterial pathogen *M. tuberculosis*, i.e. mtCA 1 (Rv1284), mtCA 2 (Rv3588c) and mtCA 3 (Rv3273). The rationale for investigating these compounds as inhibitors of these enzymes is based on the fact that many carboxylic acids strongly inhibit the growth of this pathogen by a poorly understood mechanism of action¹. Indeed, the low permeability of the mycobacterial cell wall which is due to the presence of mycolic acid-arabinogalactan complexes is regarded as a significant contributor to the intrinsic drug resistance of mycobacteria^{1,17}. This particular architecture/composition of the mycobacterial cell wall is probably also responsible for the fact that sulfonamides, poorly membrane-permeable compounds³², do not show antimycobacterial activity *in vitro* and *in vivo*¹⁷, although some of them strongly inhibit *in vitro* β -CAs present in this pathogen, some of which seem to be essential for the life cycle of the bacterium^{33,34}.

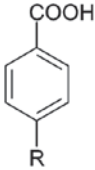
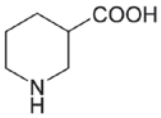
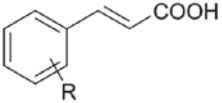
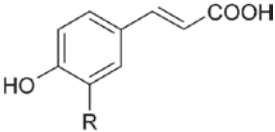
The following structure-activity relationship (SAR) can be observed from data of Table 1:

- (1) Against mtCA 1, the carboxylates **1–10** behaved as efficient micromolar inhibitors with inhibition constants in the range of 2.25–7.13 μM , comparable (but slightly better in many cases) to that of the lead compound sulfanilamide, **SA** which has a K_i of 9.84 μM (Table 1). SAR is rather straightforward with the simple substituted benzoic acids **1–5** as well as nipecotic acid **6**, showing

K_i s of 3.91–6.20 μM . The benzoic acids possessing electron withdrawing or electron donating moieties in *para* to the COOH moiety showed a similar behavior, which proves that the $\text{p}K_a$ of the carboxylic acid is not the main factor governing activity, an observation already seen for the inhibition of fungal β -CAs with aliphatic-, branched- or aromatic carboxylic acids^{35,36}. The *ortho*- and *para*-coumaric acids **7** and **8** showed the best inhibition against this isoform with K_i s of 2.25–3.03 μM . The additional OH moiety from caffeic acid **9** or the methoxy one in ferulic acid **10** lead to a roughly 2-fold decrease of the activity compared to *p*-coumaric acid **8**, demonstrating that small structural changes considerably affect the enzyme inhibitory activity.

- (2) mtCA 2 was also inhibited efficiently by carboxylates **1–10** with inhibition constants in the range of 0.59–8.10 μM (**SA** showed a K_i of 29.6 μM , being a much less effective CAI compared to the carboxylates investigated here). The aromatic carboxylates **1–4** were efficient mtCA 2 inhibitors with inhibition constants in the range of 2.12–3.67 μM , but the nitrosubstituted compound **5** (the most acidic one in the subseries) was a much weaker inhibitor with a K_i of 8.10 μM . Nipecotic acid **6** and *o*-coumaric acid **7** were also medium potency inhibitors (K_i s of 4.17–5.27 μM), whereas *p*-coumaric and caffeic acid were the best mtCA 2 inhibitors with K_i s of 0.59–1.56 μM . An extra methyl group present in **9**, as in ferulic acid **10**, led to a loss of activity, this compound showing

Table 1. Inhibition data of mycobacterial β -CA isoforms mtCA 1–3 with carboxylic acids **1–10** and sulfanilamide **SA** by a stopped-flow, CO_2 hydrase assay³⁰.

<div style="display: flex; justify-content: space-around; align-items: center;"> <div style="text-align: center;">  <p>1–5</p> </div> <div style="text-align: center;">  <p>6</p> </div> <div style="text-align: center;">  <p>7, 8</p> </div> <div style="text-align: center;">  <p>9, 10</p> </div> </div>				
Compound	R	K_i (μM)*		
		mtCA 1	mtCA 2	mtCA 3
1	Br	3.91	2.66	0.97
2	I	4.94	2.14	0.14
3	CN	6.20	2.12	0.19
4	AcNH	4.76	3.67	0.48
5	NO_2	5.87	8.10	0.52
6	–	4.89	5.27	0.11
7	2-OH	2.25	4.17	0.12
8	4-OH	3.03	1.56	0.60
9	OH	5.92	0.59	0.57
10	OMe	7.13	3.50	0.36
–	SA (sulfanilamide)	9.84	29.6	7.11

*Errors in the range of $\pm 10\%$ of the reported values, by a CO_2 hydrase assay method (from three different measurements)³⁰.

an inhibition constant of 3.50 μM (5.9 times worse than that of *p*-coumaric acid **9**).

- (3) The third bacterial CA investigated here, mtCA 3, was the most sensitive to inhibition with carboxylates **1–10**, these derivatives showing K_i s in the range of 0.11–0.97 μM , an order of magnitude better compared to that of sulfanilamide **SA** (K_i of 7.11 μM). The benzoic acids **1–5** showed a compact behavior of efficient mtCA 3 inhibitors with best compounds being those incorporating iodine and cyano moieties (**2** and **3**), whereas the least effective one was the bromobenzoic acid **1**. Nipectotic acid **6** was the best mtCA 3 inhibitor detected in this study with a K_i of 0.11 μM (Table 1) together with *o*-coumaric acid **7** (K_i of 0.12 μM). The remaining natural product carboxylates **8–10** were slightly less inhibitory compared to **7** (K_i s of 0.36–0.60 μM), although they show small structural differences with each other, proving that the inhibition of the enzyme is drastically influenced by small changes in the scaffold of the carboxylic acid.
- (4) mtCA 3 was the most prone to be inhibited by carboxylates **1–10** followed by mtCA 2, whereas mtCA 1 was the least inhibited isoform. This is quite different from the inhibition profile of the sulfonamide **SA**. Indeed, for this compound, the best inhibition was seen against mtCA 3 followed by mtCA 1 and the least inhibited isoform was mtCA 2.

One of the main problems with mtCA inhibitors investigated so far is that no inhibition of the bacterial growth *in vivo* has been observed to date (unpublished results from this laboratory and ref. Nishimori et al.¹⁷), probably due to the very hydrophilic, membrane impermeant character of sulfonamides and their difficulty to cross the mycobacteria cell wall¹. The rationale of this study was just to detect inhibitors belonging to a different chemotype, which may be derivatized easily (as amides/hydrazides) which may allow a better penetrability profile to such compounds, as for the two clinically used (pro) drugs **A** and **B** discussed above. Further studies are thus warranted to obtain highly effective and lipophilic mtCA inhibitors and prove their *in vivo* efficacy.

Conclusion

We evaluated a series of aromatic/heterocyclic carboxylates as inhibitors of three β -CAs from the bacterial pathogen *M. tuberculosis*, mtCA 1 (Rv1284), mtCA 2 (Rv3588c) and mtCA 3 (Rv3273). All three enzymes were inhibited with efficacies between the submicromolar to the micromolar one, depending on the scaffold present in the carboxylic acid. mtCA 3 was the isoform mostly inhibited by these compounds (K_i s in the range of 0.11–0.97 μM) followed by mtCA 2 (K_i s in the range of 0.59–8.10 μM), whereas against mtCA 1, these carboxylic acids showed inhibition constants in the range of 2.25–7.13 μM . This class of relatively underexplored β -CA inhibitors warrant

further *in vivo* studies, as they may have the potential for developing antimycobacterial agents with a diverse mechanism of action compared to the clinically used drugs for which many strains exhibit multi-drug or extensive multi-drug resistance.

Declaration of interest

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