



Honey as an apitherapeutic product: its inhibitory effect on urease and xanthine oxidase

Huseyin Sahin

To cite this article: Huseyin Sahin (2016) Honey as an apitherapeutic product: its inhibitory effect on urease and xanthine oxidase, Journal of Enzyme Inhibition and Medicinal Chemistry, 31:3, 490-494, DOI: [10.3109/14756366.2015.1039532](https://doi.org/10.3109/14756366.2015.1039532)

To link to this article: <https://doi.org/10.3109/14756366.2015.1039532>



Published online: 05 May 2015.



Submit your article to this journal [↗](#)



Article views: 2459



View related articles [↗](#)



View Crossmark data [↗](#)



Citing articles: 7 View citing articles [↗](#)

RESEARCH ARTICLE

Honey as an apitherapeutic product: its inhibitory effect on urease and xanthine oxidase

Huseyin Sahin

Giresun University, Espiye Vocational School, Espiye, Giresun, Turkey

Abstract

The aim of this study was to evaluate new natural inhibitor sources for the enzymes urease and xanthine oxidase (XO). Chestnut, oak and polyfloral honey extracts were used to determine inhibition effects of both enzymes. In addition to investigate inhibition, the antioxidant capacities of these honeys were determined using total phenolic content (TPC), ferric reducing antioxidant power (FRAP), and DPPH radical scavenging activity assays. Due to their high phenolic content, chestnut and oak honeys are found to be a powerful source for inhibition of both enzymes. Especially, oak honeys were efficient for urease inhibition with 0.012–0.021 g/mL IC₅₀ values, and also chestnut honeys were powerful for XO inhibition with 0.028–0.039 g/mL IC₅₀ values. Regular daily consumption of these honeys can prevent gastric ulcers deriving from *Helicobacter pylori* and pathological disorders mediated by reactive oxygen species.

Keywords

Antioxidant, chestnut, oak, urease, xanthine oxidase

History

Received 9 March 2015
Revised 14 March 2015
Accepted 17 March 2015
Published online 5 May 2015

Introduction

Urease is an enzyme that catalyzes the hydrolysis of urea into carbon dioxide and ammonia¹. Levels of this enzyme, which is responsible for reducing urea accumulation, are also involved in the development of urolithiasis, pyelonephritis, hepatic encephalopathy, hepatic coma urolithiasis, and urinary catheter encrustation in humans and animals². Urease activity is essential for buffering the acidic pH value in the stomach, nutrient acquisition and improving the ability of *Helicobacter pylori* to colonize the gastric epithelium³. Its inhibition is very important for the treatment of *H. pylori*-related diseases.

In addition to this direct potential disease risk, reactive oxygen species (ROS), which are formed through various pathways, are also significant risk factors for diseases in different systems⁴. ROS are involved in oxidative damage to lipids, proteins, and nucleic acids. In general, ROS are formed through both physiological and non-physiological pathways. Some of the enzymes of myeloperoxidase, aldehyde oxidase, nitric oxide synthase and xanthine oxidase (XO) catalyze the formation of some ROS^{5,6}. Many ROS are generated by XO to catalyze the oxidation of hypoxanthine into xanthine^{7,8}. XO is responsible for oxidative damage that causes many pathological diseases, such as gout, hyperuricemia, hepatitis, carcinogenesis, and aging^{9,10}. The regulation of XO is an important means of preventing inflammation⁶. In particular, increased XO expression can result in significant vascular endothelium damage¹¹, as well as atherosclerosis. XO inhibitor scanning can prevent the development of endothelial dysfunction and atherosclerosis^{11,12}. Some chemicals, such as allopurinol⁸ and

pyrazoles¹³, have been used for only clinically XO inhibitor. Besides these chemicals, a number of natural compounds, such as caffeic acid^{11,14,15}, rutin¹⁶, and chestnut honey¹⁷, have been reported to inhibit XO, and foods rich in phenolic compounds are recommended for reducing blood concentrations of uric acid in gout.

Honey is one of these natural products, and is rich in phenolics as antioxidant. The antioxidant capacity of honey is affected by several factors, such as the floral source involved, and seasonal, geographical and environmental conditions^{18,19}.

Nasuti et al.¹⁷ and Can et al.¹⁹ investigated the biological properties of some honeys. They reported dark honeys chestnut and oak possessing higher levels of phenolic compounds and antioxidant capacities, and greater apitherapeutic functions. Another finding of these studies is that phenolic acids and flavonoids in honey could be used as a vital enzyme inhibitors^{17,19}. According to another aspect, the use of natural products, rather than chemical-based drugs, is desirable in the inhibition of these enzymes^{3,20,21}. Recent studies have revealed that natural products are effective in the elimination of ROS and as enzyme inhibitors^{22–24}. Considering these information, the objective of this study is to determine the antioxidant properties and inhibition effects of different types of honey (oak, chestnut and polyfloral honey) on urease and XO for the first time.

Materials and methods

Reagents

Enzymes and their substrates were supplied in the form of bovine milk XO -xanthine and jack bean urease-urea by Sigma-Aldrich (St. Louis, MO). Hydrochloric acid (HCl), glacial acetic acid and Folin–Ciocalteu reagent were obtained from LiChrosolv® (Merck KGaA, Darmstadt, Germany). High quality ultra-pure water was supplied by Human Zeneer Navi Power I Integrate

Address for correspondence: Huseyin Sahin, PhD, Giresun University, Espiye Vocational School, Espiye 28600, Giresun, Turkey. Tel: +904546116007. Fax: +904546116008. E-mail: huseyin.sahin@giresun.edu.tr

(Human Corporation, Seoul, South Korea). Potassium phosphate dibasic (K_2HPO_4), ethylenediaminetetraacetic acid (EDTA), lithium chloride (LiCl), phenol, sodium nitroprusside, sodium hydroxide (NaOH), sodium hypochlorite (NaOCl), sodium phosphate dibasic (Na_2HPO_4), sodium phosphate monobasic monohydrate ($NaH_2PO_4 \cdot H_2O$), gallic acid, sodium carbonate (Na_2CO_3), sodium acetate trihydrate ($NaCH_3COO \cdot 3H_2O$), iron(III) chloride hexahydrate ($FeCl_3 \cdot 6H_2O$), 2,4,6-Tris(2-pyridyl)-s-triazine (TPTZ), iron(II) sulfate heptahydrate ($FeSO_4 \cdot 7H_2O$), and DPPH (2,2-Diphenyl-1-(2,4,6-trinitrophenyl) hydrazyl) were purchased from Sigma-Aldrich (St. Louis, MO). LC syringe filters (RC-membrane, 0.2 μ m) were obtained from Sartorius Minisart RC 15, Sartorius (Darmstadt, Germany).

Honey samples

The samples were collected by experienced beekeepers in the 2014 harvest season. Three varieties of honey, chestnut (*Castanea sativa* Mill.), oak (*Quercus robur* L.), and polyfloral honey were used in this study. For tagging, melissopalynological analysis was performed following the method described by Louveaux et al.²⁵. Acetolyzed pollen grains were mounted on glycerin jelly and sealed with paraffin. In order to determine the pollen-type frequency classes, 500 pollen grains were counted and classified in terms of dominant pollen (more than 45%). Pollen analyses and honey properties are given in Table 1.

Honey preparation

Approximately 5 g of honey sample was extracted with 20-mL distilled water in a flask attached to a condenser at 60 °C, over 6 h. The extract was subsequently filtered to remove particles, and the final volume was adjusted with distilled water.

Urease inhibition assay

The urease inhibiting activity of the aquatic honey samples was determined by measuring ammonia production using the indophenol method²⁶. Briefly, reaction mixtures including 200- μ L jack bean urease, 500 μ L of buffer (100 mM urea, 0.01 M K_2HPO_4 , 1 mM EDTA and 0.01 M LiCl, pH 8.2) and 100 μ L honey extract samples were incubated at room temperature for 20 min. Phenol reagent (550 μ L, 1% w/v phenol and 0.005% w/v sodium nitroprusside) and alkali reagent (650 μ L, 0.5% w/v sodium hydroxide and 0.1% v/v NaOCl) were added, and the increasing absorbance at 625 nm was measured after 50 min, using a Thermo Scientific Evolution 260 spectrophotometer (Thermo Scientific, Waltham, MA). The IC_{50} value was determined as the concentration of sample causing 50% inhibition of maximal activity.

In vitro anti-xanthine oxidase assay

The XO inhibitory activity of honey extracts was determined using the UV spectroscopy technique at 295 nm²⁷ with some slight modifications. The reaction mixture consisted of 0.5 mL of the test compound, 0.77 mL of phosphate buffer (pH 7.8) and 0.07 mL of bovine milk XO prepared immediately before use. After pre-incubation at 25 °C for 15 min, the reaction was initiated by the addition 0.66 mL of substrate solution into the mixture. The assay mixture was then incubated at 25 °C for 15 min. The reaction was stopped by adding 0.2 mL of 0.5 N HCl, and the absorbance was measured at 295 nm using a Thermo Scientific Evolution 260 spectrophotometer (Thermo Scientific, Waltham, MA). The half-maximal inhibitory concentration (IC_{50}) was generated from the absorbance data.

Determination of antioxidant capacity

The antioxidant capacities of the honey samples were determined using three different assays; TPC, ferric reducing antioxidant power and free radical scavenging activity of DPPH.

TPCs were determined using the Folin–Ciocalteu procedure with gallic acid as standard²⁸. Briefly, 20 μ L samples (1 mg/mL), 400 μ L of 0.2 N Folin–Ciocalteu reagent and 680 μ L of distilled water were mixed, and the mixture was vortexed. Following 3-min incubation, 400 μ L of Na_2CO_3 (10%) solution was added and vortexed. After vortexing, the mixture was incubated with intermittent shaking for 2 h at 25 °C. Absorbance was measured at 760 nm at the end of the incubation period. TPC concentration was calculated as mg of gallic acid equivalents per gram sample, using a standard graph.

Working FRAP reagent was obtained as required by mixing 25 mL acetate buffer (300 mM, pH 3.6), with 2.5 mL of 10 mM TPTZ solution in 40 mM HCl and 2.5 mL of 20 mM $FeCl_3 \cdot 6H_2O$ solution²⁹. Next, 100 μ L of the honey sample was mixed with 3 mL of freshly prepared FRAP reagent and incubated for 4 min at room temperature. Absorbance was read at 593 nm against reagent blank. $FeSO_4 \cdot 7H_2O$ was used a positive control to construct a reference curve (31.25–1000 μ M, $r^2 = 0.999$), FRAP values were expressed as μ mol $FeSO_4 \cdot 7H_2O$ equivalent of g sample.

The scavenging of 2,2-diphenyl-1-picrylhydrazyl (DPPH) radicals was used to determine the radical scavenging activity of the honey samples. The colorimetric test was performed using the Molyneux method³⁰. For each sample, various concentrations of 0.75 mL of extracts of honey were mixed with 0.75 mL of 0.1 mM of DPPH in methanol. Radical scavenging activity was measured using Trolox as standard, and the values were expressed as SC_{50} (mg sample per mL), the concentration of samples causing 50% scavenging of DPPH radicals.

Table 1. Data from studies included in the identification markers.

Honey code	Flora	Location	Dom.pollen%	Major familia	Predominant pollen
H1	Chestnut	Zonguldak	70	<i>Fagaceae</i>	<i>Castaneasativa</i> Mill.
H2	Chestnut	Trabzon	68	<i>Fagaceae</i>	<i>Castaneasativa</i> Mill.
H3	Chestnut	Giresun	66	<i>Fagaceae</i>	<i>Castaneasativa</i> Mill.
H4	Chestnut	Rize	72	<i>Fagaceae</i>	<i>Castaneasativa</i> Mill.
H5	Oak	Kırklareli	62	<i>Fagaceae</i>	<i>Quercusrobur</i> L.
H6	Oak	Kırklareli	52	<i>Fagaceae</i>	<i>Quercusrobur</i> L.
H7	Oak	Kırklareli	60	<i>Fagaceae</i>	<i>Quercusrobur</i> L.
H8	Oak	Bolu	55	<i>Fagaceae</i>	<i>Quercusrobur</i> L.
H9	Polyfloral	Aydın	N.D.	<i>Verbenaceae, Fagaceae, Rosaveae, Lamiaceae, Fabaceae, Rutaceae</i>	*
H10	Polyfloral	Bolu	N.D.	<i>Malvaceae, Fabaceae, Fagaceae asteraceae, Alliaceae</i>	*
H11	Polyfloral	Giresun	N.D.	<i>Fagaceae, Ericaveae, Rocaceae, Fabaceae, Tiliaceae</i>	*
H12	Polyfloral	Kars	N.D.	<i>Fabaceae, Lamiaceae, Betulaceae, Apiaceae, Campanulaceae, Papaveraceae</i>	*

ND, not detected.

*Dominant pollen was not present.

Table 2. IC₅₀ values of each enzyme and antioxidant properties of honey samples*.

Honey types	Honey code	Urease-IC ₅₀ (g/mL)	Xanthine oxidase-IC ₅₀ (g/mL)	Total phenolics (mg GAE/100 g sample)	FRAP (μmol FeSO ₄ ·7H ₂ O/g sample)	DPPH-SC ₅₀ (mg/mL)
Chestnut	H1	0.034 ± 0.001 ^g	0.039 ± 0.000 ^a	41.170 ± 0.412 ^e	3.545 ± 0.036 ^f	40.866 ± 0.409 ^d
	H2	0.010 ± 0.000 ^a	0.028 ± 0.000 ^a	52.672 ± 1.053 ^g	4.336 ± 0.087 ^h	41.350 ± 0.827 ^d
	H3	0.034 ± 0.001 ^g	0.033 ± 0.000 ^a	38.900 ± 0.960 ^d	3.111 ± 0.078 ^e	40.874 ± 1.022 ^d
	H4	0.025 ± 0.000 ^f	0.030 ± 0.000 ^a	65.300 ± 1.306 ⁱ	4.690 ± 0.094 ⁱ	35.100 ± 0.702 ^{c,d}
Oak	H5	0.012 ± 0.000 ^b	0.101 ± 0.002 ^b	62.260 ± 0.623 ^h	4.247 ± 0.085 ^h	18.350 ± 0.367 ^b
	H6	0.014 ± 0.000 ^c	0.110 ± 0.001 ^{b,c}	57.674 ± 1.154 ^g	4.669 ± 0.117 ⁱ	10.892 ± 0.272 ^a
	H7	0.021 ± 0.000 ^c	0.147 ± 0.004 ^c	36.806 ± 0.972 ^d	3.104 ± 0.062 ^e	38.160 ± 0.763 ^{c,d}
	H8	0.018 ± 0.000 ^d	0.135 ± 0.010 ^{b,c}	45.460 ± 0.909 ^f	3.680 ± 0.037 ^g	32.100 ± 0.321 ^c
Polyfloral	H9	0.045 ± 0.001 ^h	0.384 ± 0.004 ^e	10.193 ± 0.204 ^a	0.990 ± 0.025 ^b	250.555 ± 6.263 ^g
	H10	0.052 ± 0.001 ⁱ	0.255 ± 0.006 ^d	29.480 ± 0.737 ^c	1.895 ± 0.038 ^d	76.153 ± 1.523 ^e
	H11	0.069 ± 0.001 ^j	0.452 ± 0.005 ^f	9.400 ± 0.188 ^a	0.527 ± 0.013 ^a	411.416 ± 10.285 ^h
	H12	0.076 ± 0.001 ^k	0.361 ± 0.083 ^e	19.500 ± 0.390 ^b	1.440 ± 0.014 ^c	152.200 ± 1.522 ^f

*Means ± standard deviations; Different letters (a–k) in the same columns are significantly different at the 5% level ($p < 0.05$)

Table 3. “Paired samples-*t*” test correlation coefficients.

	Paired differences	Urease-IC ₅₀	Xanthine oxidase-IC ₅₀	Total phenolics	FRAP	DPPH-SC ₅₀
Urease-IC ₅₀	Pearson Correlation	1.000	0.804 [†]	−0.839 [†]	−0.881 [†]	0.741 [†]
	Sig. (two-tailed)		0.000	0.000	0.000	0.000
	<i>N</i>	36.000	36.000	36.000	36.000	36.000
Xanthine oxidase-IC ₅₀	Pearson Correlation	0.804 [†]	1.000	−0.863 [†]	−0.907 [†]	0.864 [†]
	Sig. (two-tailed)	0.000		0.000	0.000	0.000
	<i>N</i>	36.000	36.000	36.000	36.000	36.000
Total phenolics	Pearson Correlation	−0.839 [†]	−0.863 [†]	1.000	0.979 [†]	−0.826 [†]
	Sig. (two-tailed)	0.000	0.000		0.000	0.000
	<i>N</i>	36.000	36.000	36.000	36.000	36.000
FRAP	Pearson Correlation	−0.881 [†]	−0.907 [†]	0.979 [†]	1.000	−0.851 [†]
	Sig. (two-tailed)	0.000	0.000	0.000		0.000
	<i>N</i>	36.000	36.000	36.000	36.000	36.000
DPPH-SC ₅₀	Pearson Correlation	0.741 [†]	0.864 [†]	−0.826 [†]	−0.851 [†]	1.000
	Sig. (two-tailed)	0.000	0.000	0.000	0.000	
	<i>N</i>	36.000	36.000	36.000	36.000	36.000

[†]Correlation is significant at the 0.01 level (two-tailed).

Statistical analysis

The results were given in the form of arithmetical mean values and standard deviations. The SPSS 13.00 for Windows software package was used for the statistical analysis of the gathered data (SPSS Inc., Chicago, IL). Significance of the analysis of the results was based on the Kruskal–Wallis test and Pearson correlation. Significant differences were statistically considered at the level of $p < 0.05$.

Results and discussion

The purpose of this study was to determine whether honey exhibits an inhibitory effect on two anti-inflammatory enzymes of vital importance to human health. Common inhibitors of both enzymes will be important to alternative medicine as protective agents and in the treatment of gastric ulcer and gout. IC₅₀ values for each enzyme in the analyzed honey samples are given in Table 2.

Before examining enzyme inhibition, the antioxidant capacities of the honeys used as inhibitor sources were clarified. Three different methods were used to evaluate the antioxidant capacity of the honeys; TPC, the ferric reducing antioxidant assay (FRAP) reflecting total antioxidant capacity and the DPPH assay showing total radical scavenging capacity. TPC of the honeys varied widely, from 9.400 to 65.000 mg GAE/100 g sample (Table 2).

Oak and chestnut honeys had higher level of TPC concentrations which showed the statistically significant differences from each other, polyfloral honeys exhibited the lowest TPC. A positive correlation was determined between TPC and FRAP values (r^2 : 0.979, $p < 0.01$), and between TPC and DPPH activity (r^2 : −0.826, $p < 0.01$) (Table 3). Higher TPC indicates higher antioxidant capacity, as well as DPPH radical scavenging activities.

Previous studies have reported that chestnut and oak honeys are both dark colored and have higher phenolic contents and associated antioxidant capacity than light-colored honeys^{17,19,31}. Seventy to eighty percent of the dry weight of honey consists of carbohydrate and 2% of secondary metabolites. The bioactivity of honey derives mainly from these secondary metabolite agents in its structure, that varies depending on the floral sources involved^{32,33}. Phenolic compounds, vitamins (A, E and C), free amino acids, proteins and enzymes represent the secondary metabolites of honey. The great majority of these compounds that give rise to the true quality of honey are phenolic structure molecules, phenolic acids, flavanols, pro-anthocyanins and tannins that determine the aroma, taste and sensory characteristics of honey^{34,35}. These secondary metabolites possess not only anti-oxidant activities, but also anti-microbial, anti-tumoral, and anti-inflammatory functions^{36–38}.

All three honeys in the study inhibited urease and XO in a manner dependent on the varying IC₅₀ values and concentrations. Oak honey exhibited the highest degree of inhibition of urease, followed by chestnut honey and polyfloral honey. The inhibitory value of oak honey was approximately 3 times greater than that of polyfloral honey.

The high urease inhibitor activity of oak honey may derive from its high phenolic composition. Indeed, a negative correlation was determined between TPC and urease enzyme inhibition (r^2 : -0.839, $p < 0.01$) (Table 3). This significantly high correlation derives from the presence of phenolic compounds in oak honey. A previous study of ours reported that oak honey is rich in rutin, gallic acid and protocatechuic acid¹⁹. Also, Modolo et al.³⁹ stated that some flavonoids as rutin, quercetin, and luteolin are effective in the *H. pylori* inhibition with IC₅₀ values of 11.2 µM, 67.6 µM, and 35.5 µM, respectively. In another study, the IC₅₀ values of *Eucalyptus grandis* stem and bark methanolic extracts ranged between 6.5 and 50.0 mg/mL⁴⁰.

Similarly to urease, a negative correlation was also determined between honey TPC levels and XO inhibitions (r^2 : -0.863, $p < 0.01$) (Table 3). Compared to oak honey, chestnut had a low inhibitory effect on XO and exhibited lower IC₅₀ inhibition values. A previous study of ours reported that chestnut honey is rich in quercetin, caffeic acid, coumaric acid and protocatechuic acid¹⁹. One of the study reported that caffeic acid and its species are a significant inhibitor for XO⁶, and another reported that chestnut honey exhibits XO inhibition¹⁷.

In conclusion, honeys studied in this research particularly with their high levels of TPC may be considered as a good nutrition source for the inhibition of the enzymes urease and XO. Regular consumption of these products may contribute to a reduction in several forms of ROS-mediated pathological injury.

Declaration of interest

This study was supported by the KTU-BAP Project (Project number: 970). The author reports no financial or other conflict of interest relevant to the subject matter of this article.

References

- Cui Y-M, Dong X-W, Chen W, et al. Synthesis, inhibitory activity and molecular docking studies of two Cu(II) complexes against *Helicobacter pylori* urease. *J Enzyme Inhib Med Chem* 2012;27: 528–32.
- Li X, Mobley HLT. Vaccines for *Proteus mirabilis* in urinary tract infection. *Int J Antimicrob Agents* 2002;19:461–5.
- Chi-Hui Cho WC, Chandwick, VS. Novel anti-*Helicobacter pylori* agents. *Exp Opin Ther Patents* 2000;10:1221–32.
- Sasidharan I, Sundaresan A, Nisha VM, et al. Inhibitory effect of *Terminalia chebula* Retz. fruit extracts on digestive enzyme related to diabetes and oxidative stress. *J Enzyme Inhib Med Chem* 2012; 27:578–86.
- Weidert ER, Schoenborn SO, Cantu-Medellin N, et al. Inhibition of xanthine oxidase by the aldehyde oxidase inhibitor raloxifene: implications for identifying molybdopterin nitrite reductases. *Nitric Oxide* 2014;37:41–5.
- Chang CY, Lee FW, Chen CS, et al. Structure-activity relationship of C6-C3 phenylpropanoids on xanthine oxidase-inhibiting and free radical-scavenging activities. *Free Radic Biol Med* 2007;43: 1541–51.
- Pauff, JM, Hille R. Inhibition studies of bovine xanthine oxidase by luteolin, silibinin, quercetin, and curcumin. *J Nat Prod* 2009;72: 725–31.
- Ryu HW, Lee JH, Kang JE, et al. Inhibition of xanthine oxidase by phenolic phytochemicals from *Broussonetia papyrifera*. *J Korean Soc Appl Biol Chem* 2012;55:587–94.
- Mehta SK, Nayeem N. Natural xanthine oxidase inhibitors for management of gout: a review. *RRJMHS* 2014;3:1–13.
- Bandgar BP, Jalde SS, Korbadi BL, et al. Synthesis and antioxidant, cytotoxicity and antimicrobial activities of novel curcumin mimics. *J Enzyme Inhib Med Chem* 2012;27:267–74.
- Schröder K, Vecchione C, Jung O, et al. Xanthine oxidase inhibitor tungsten prevents the development of atherosclerosis in ApoE knockout mice fed a Western-type diet. *Free Radic Biol Med* 2006; 41:1353–60.
- Nomura J, Busso N, Ives A, et al. Xanthine oxidase inhibition by febuxostat attenuates experimental atherosclerosis in mice. *Sci Rep* 2014;4:1–9.
- Allouche F, Chabchoub F, Carta F, Supuran CT. Synthesis of aminocyanopyrazoles via a multi-component reaction and anti-carbonic anhydrase inhibitory activity of their sulfamide derivatives against cytosolic and transmembrane isoforms. *J Enzyme Inhib Med Chem* 2013;28:343–9.
- Russo A, Longo R, Vanella A. Antioxidant activity of propolis: role of caffeic acid phenethyl ester and galangin. *Fitoterapia* 2002;73: 21–9.
- Masuda T, Shingai Y, Takahashi C, et al. Identification of a potent xanthine oxidase inhibitor from oxidation of caffeic acid. *Free Radic Biol Med* 2014;69:300–7.
- Takahama U, Koga Y, Hirota S, Yamauchi R. Inhibition of xanthine oxidase activity by an oxathiolanone derivative of quercetin. *Food Chem* 2011;126:1808–11.
- Nasuti C, Gabbianelli R, Falcioni G, Cantalamessa F. Antioxidative and gastroprotective activities of anti-inflammatory formulations derived from chestnut honey in rats. *Nutr Res* 2006;26:130–7.
- Sahin H, Aliyazıcıoğlu R, Yıldız O, et al. Honey, pollen, and propolis extracts show potent inhibitory activity against the zinc metalloenzyme carbonic anhydrase. *J Enzyme Inhib Med Chem* 2011;26:440–4.
- Can Z, Yildiz O, Sahin, H, et al. An investigation of Turkish honeys: their physico-chemical properties, antioxidant capacities and phenolic profiles. *Food Chem* 2015;180:133–41.
- Paun G, Litescu SC, Neagu E, et al. Evaluation of *Geranium* spp., *Helleborus* spp. and *Hyssopus* spp. polyphenolic extracts inhibitory activity against urease and α -chymotrypsin. *J Enzyme Inhib Med Chem* 2014;29:28–34.
- Ndemangou B, Sielinou VT, Vardamides JC, et al. Urease inhibitory isoflavonoids from different parts of *Calopogonium mucunoides* (Fabaceae). *J Enzyme Inhib Med Chem* 2013;28:1156–61.
- Eteraf-Oskouei T, Najafi M. Traditional and modern uses of natural honey in human diseases: a review. *Iran J Basic Med Sci* 2013;16: 731–42.
- Matongo F, Nwodo UU. *In vitro* assessment of *Helicobacter pylori* ureases inhibition by honey fractions. *Arch Med Res* 2014; 45:540–6.
- Ekinci D, Kurbanoglu NI, Salamci E, et al. Carbonic anhydrase inhibitors: inhibition of human and bovine isoenzymes by benzenesulphonamides, cyclitols and phenolic compounds. *J Enzyme Inhib Med Chem* 2012;27:845–8.
- Louveaux J, Maurizio A, Vorwohl G. Methods of melissopalynology. *Bee World* 1978;59:139–57.
- Weatherburn MW. Phenol-hypochlorite reaction for determination of ammonia. *Anal Chem* 1967;39:971–4.
- Van Hoorn DEC, Nijveldt RJ, Van Leeuwen PAM, et al. Accurate prediction of xanthine oxidase inhibition based on the structure of flavonoids. *Eur J Pharmacol* 2002;451:111–18.
- Singleton VL, Rossi JL. Colorimetry of total phenolics with phosphomolybdic phosphotungstic acid reagents. *Am J Enol Viticult* 1965;16:144–58.
- Benzie IFF, Strain JJ. Ferric reducing/antioxidant power assay: direct measure of total antioxidant activity of biological fluids and modified version for simultaneous measurement of total antioxidant power and ascorbic acid concentration. *Method Enzymol* 1999;299: 15–27.
- Molyneux P. The use of the stable free radical diphenylpicrylhydrazyl (DPPH) for estimating antioxidant activity. *Songklanakarin J Sci Technol* 2004;26:211–19.
- Tezcan F, Kolaylı S, Sahin H, et al. Evaluation of organic acid, saccharide composition and antioxidant properties of some authentic Turkish honeys. *J Food Nutr Res* 2011;50:33–40.
- Küçük M, Kolaylı S, Karaoğlu Ş, et al. Biological activities and chemical composition of three honeys of different types from Anatolia. *Food Chem* 2007;100:526–34.

33. Sahin H, Can Z, Yildiz O, et al. Inhibition of carbonic anhydrase isozymes I and II with natural products extracted from plants, mushrooms and honey. *J Enzyme Inhib Med Chem* 2012;27: 395–402.
34. Kolaylı S, Yildiz O, Sahin H, Aliyazıcıoğlu R. Biochemistry and physicochemical properties of honey. In: Boukraa L, ed. *Honey in traditional and modern medicine*. Boca Raton: CRC Press, Taylor & Francis Group; 2013:21–35.
35. Yıldız O, Sahin H, Kolaylı S. Culinary uses of honey. In: Boukraa L, ed. *Honey in traditional and modern medicine*. Boca Raton: CRC Press, Taylor & Francis Group; 2013:435–40.
36. Yıldız O, Karahalil F, Can Z, et al. Total monoamine oxidase (MAO) inhibition by chestnut honey, pollen and propolis. *J Enzyme Inhib Med Chem* 2014;29:690–4.
37. Yıldız O, Can Z, Saral O, et al. Hepatoprotective potential of chestnut bee pollen on carbon tetrachloride-induced hepatic damages in rats. *J Evid Based Complementary Altern Med* 2013;461478: 1–9.
38. Jeong EJ, Seo H, Yang H, et al. Anti-inflammatory phenolics isolated from *Juniperus rigida* leaves and twigs in lipopolysaccharide-stimulated RAW264.7 macrophage cells. *J Enzyme Inhib Med Chem* 2012;27:875–9.
39. Modolo LV, de Souza AX, Horta LP, et al. An overview on the potential of natural products as ureases inhibitors: a review. *J Adv Res* 2015;6:35–44.
40. Adeniyi BA, Lawal TO, Mahady GB. *In vitro* susceptibility of *Helicobacter pylori* to extracts of *Eucalyptus camaldulensis* and *Eucalyptus torelliana*. *Pharm Biol* 2009;47:99–102.